

CrossMark  
click for updatesCite this: *J. Mater. Chem. A*, 2015, 3, 11768Received 10th March 2015  
Accepted 25th April 2015

DOI: 10.1039/c5ta01778e

www.rsc.org/MaterialsA

## Enhancement of thermoelectric performance of $\beta$ -Zn<sub>4</sub>Sb<sub>3</sub> through resonant distortion of electronic density of states doped with Gd<sup>†</sup>

Baojin Ren,<sup>a</sup> Mian Liu,<sup>a</sup> Xiaoguang Li,<sup>b,c</sup> Xiaoying Qin,<sup>\*a</sup> Di Li,<sup>a</sup> Tianhua Zou,<sup>a</sup> Guolong Sun,<sup>a</sup> Yuanyue Li,<sup>a</sup> Hongxing Xin<sup>a</sup> and Jian Zhang<sup>a</sup>

The thermoelectric properties of Gd-doped  $\beta$ -Zn<sub>4</sub>Sb<sub>3</sub> are investigated. The results indicate that Gd-doping not only causes a 41  $\mu\text{V K}^{-1}$  increase in thermopower owing to resonant distortion of DOS but also results in  $\sim 15\%$  reduction in thermal conductivity at a doping content of 0.2%. Consequently, a largest value of  $ZT = 1.2$  is achieved at 655 K.

### Introduction

Recently, thermoelectric materials have attracted a great deal of attention for their possible applications in energy conversion and power generation.<sup>1–5</sup> However, their applications are limited by the relatively low conversion efficiency, which is quantified by thermoelectric figure of merit,  $ZT = (S^2/\rho\kappa)T$ , where  $\rho$ ,  $S$  and  $\kappa$  are the electrical resistivity, the thermopower, and the total thermal conductivity.  $\beta$ -Zn<sub>4</sub>Sb<sub>3</sub> is a very potential thermoelectric material in the moderate temperature range because it possesses high thermoelectric performance<sup>6</sup> and is made of relatively cheap and nontoxic elements. In order to improve its thermoelectric properties, a doping approach was used to optimize its carrier concentration and reduce its thermal conductivity. For instance, doping of Pb, Bi, Nb, Cu, Cd, Co, Sn, In, Al, Ga, Mg, Ge and Hg<sup>7–19</sup> have been investigated so far. The results showed that, however, only a small amount doping of one of these elements could lead to limited improvement of its thermoelectric performance. The main reasons would lie in these factors: (i) thermal conductivity of  $\beta$ -Zn<sub>4</sub>Sb<sub>3</sub> is very low ( $< 1 \text{ W m}^{-1} \text{ K}^{-1}$ )<sup>6,20</sup> which is close to the lower limit for the thermal conductivity in solids; (ii) the hole concentration of pristine  $\beta$ -Zn<sub>4</sub>Sb<sub>3</sub> is already close to the

optimum (in the order of  $10^{18}$  to  $10^{19} \text{ cm}^{-3}$ ).<sup>6,16</sup> These characters of  $\beta$ -Zn<sub>4</sub>Sb<sub>3</sub> suggest that it is difficult to enhance its thermoelectric performance through conventional doping, unless thermopower  $S$  can be extra elevated upon doping.

Recently, Heremans *et al.*<sup>21</sup> found that doping can give rise to resonant distortion of electronic density of states through the use of the thallium impurity levels in PbTe, resulting in enhancement of its  $ZT$ . According to the Mahan Sofo theory,<sup>22</sup> the local increase in the density of states  $g(E)$  (DOS) can strengthen thermopower ( $S$ ). The effect of this local increase in DOS on  $S$  is given by the Mott expression (eqn (1)).

$$S = \frac{\pi^2}{3} \frac{k_B}{q} k_B T \left\{ \frac{d[\ln(\sigma(E))]}{dE} \right\}_{E=E_F} \quad (1)$$
$$= \frac{\pi^2}{3} \frac{k_B}{T} k_B T \left\{ \frac{1}{p} \frac{dp(E)}{dE} + \frac{1}{\mu} \frac{d\mu(E)}{dE} \right\}_{E=E_F}$$

Here,  $S$  depends on the energy-dependent electrical conductivity  $\sigma(E) = p(E)q\mu(E)$  taken at the Fermi energy  $E_F$ ,<sup>23</sup> with  $p(E) = g(E)f(E)$ , where  $p(E)$  is the carrier concentration,  $f(E)$  is the Fermi function,  $q$  the carrier charge, and  $\mu(E)$  the mobility. Eqn (1) shows that an increased energy dependence of  $p(E)$ , for instance by a local increase in DOS  $g(E)$ , can enhance thermopower. Very recently, our group<sup>24–26</sup> found that rare earth element doping can significantly improve thermopower of  $\beta$ -Zn<sub>4</sub>Sb<sub>3</sub>, which could be ascribed to the occurrence of the resonant distortion of DOS.

In the present work, rare earth Gd was chosen as a dopant to explore possible resonant distortion of DOS in  $\beta$ -Zn<sub>4</sub>Sb<sub>3</sub>. Our results show that besides large reduction of thermal conductivity  $\kappa$ , thermopower  $S$  of Gd doped compound  $\beta$ -(Zn<sub>1-x</sub>Gd<sub>x</sub>)<sub>4</sub>Sb<sub>3</sub> increases by  $\sim 40 \mu\text{V K}^{-1}$  as  $x = 0.002$  and  $0.003$ ; meanwhile, density-of-states effective mass of  $\beta$ -Zn<sub>4</sub>Sb<sub>3</sub> is estimated to be around 1.3–1.7 times larger than that of the un-doped one, signifying the occurrence of resonant distortion of DOS, which is verified by low-temperature ( $< 4 \text{ K}$ ) specific heat. Our first-principles calculations further reveals that the resonant distortion near edge of the valence band originates mainly from

<sup>a</sup>Key Laboratory of Materials Physics, Institute of Solid State Physics, Chinese Academy of Science, 230031 Hefei, P. R. China. E-mail: xyqin@issp.ac.cn; Fax: +86 0551 65591434; Tel: +86 0551 65592750

<sup>b</sup>Hefei National Laboratory for Physical Sciences at Microscale, Department of Physics, University of Science and Technology of China, 230026 Hefei, P. R. China

<sup>c</sup>Collaborative Innovation Center of Advanced Microstructures, Nanjing University, Nanjing 210093, China

<sup>†</sup> Electronic supplementary information (ESI) available. See DOI: 10.1039/c5ta01778e

contribution of d-electrons of Gd. Due to both enhanced  $S$  and reduced  $\kappa$ , a largest value of  $ZT$  of 1.2 (at 655 K) is achieved for  $\beta$ -(Zn<sub>1-x</sub>Gd<sub>x</sub>)<sub>4</sub>Sb<sub>3</sub> ( $x = 0.002$ ) at 655 K, which is around 1.6 times larger than that of the un-doped sample.

## Experimental procedures

$\beta$ -(Zn<sub>1-x</sub>Gd<sub>x</sub>)<sub>4</sub>Sb<sub>3</sub> ( $x = 0, 0.001, 0.002$  and  $0.003$ ) compounds were synthesized by the melting and hot-pressing method ( $\sim 95\%$  theoretical density). The phase structures of the obtained samples were determined by using X-ray diffraction (XRD) with Cu K $\alpha$  radiation ( $\lambda = 1.540598 \text{ \AA}$ ). Accuracy lattice parameters were measured with an XRD by using a Si standard for calibration. The thermal conductivity  $\kappa$  was calculated using the relationship  $\kappa = \alpha C_p \rho$ , in which thermal diffusivity  $\alpha$  measured by the laser flash method (LFA 457) in an Ar atmosphere from 300 K to 655 K, the heat capacity  $C_p$  was measured with a Perkin Elmer Diamond DSC, and density  $\rho$  was measured in ethanol by using Archimedes' method. The electrical resistivity and thermopower were measured simultaneously using the standard four-probe method (ZEM-3) in a He atmosphere from 300 K to 655 K. The Hall coefficients and low-temperature heat capacity were measured by using a physical property measurement system (PPMS, Quantum Design) at 300 K and 2–4 K, respectively.

## Computational methods

Although  $\beta$ -Zn<sub>4</sub>Sb<sub>3</sub> was already discovered and studied for decades, a complete view on its true crystal structure was presented only recently.<sup>27–29</sup> Cargnoni *et al.*<sup>30</sup> had proposed a recognized model that consists of three types of structures to simplify the practical structure, it was noted that the three types of structures have very similar DOS regardless of the zinc content and structure.<sup>31</sup> Hence, for simplicity, we utilized the crystal structure of a hypothetical disorder-free  $\beta$ -Zn<sub>4</sub>Sb<sub>3</sub> with a framework of Zn<sub>36</sub>Sb<sub>30</sub>, one of the basic structures in Cargnoni's model. Then *ab initio* electrical structure calculations were carried out for the Zn-substituted compounds GdZn<sub>35</sub>Sb<sub>30</sub>.

Our calculations are performed within the framework of the density-functional theory, with the PBE generalized gradient approximation to the exchange correlation energy, and the valence electron interaction was modeled by the projector augmented wave potential, as implemented in the Vienna *ab initio* simulation package (VASP).<sup>32–34</sup> The plane wave cut off and

$k$ -point density were obtained using the Monkhorst–Pack method. Structural relaxations have been performed by using the conjugate gradient algorithm. The ionic coordinates and the unit cell's size and shape were optimized simultaneously to eliminate structures with internal stress.

## Results and discussion

The phase of  $\beta$ -(Zn<sub>1-x</sub>Gd<sub>x</sub>)<sub>4</sub>Sb<sub>3</sub> ( $x = 0, 0.001, 0.002$  and  $0.003$ ) is analyzed with an XRD at room temperature. All diffraction peaks perfectly correspond to the  $\beta$ -Zn<sub>4</sub>Sb<sub>3</sub> (JCPD no. 89-1969) phase, no obvious impurity phase being detected. With increasing Gd content  $x$  from 0 to 0.003, the lattice constants  $a$  and  $c$  increase monotonically as shown in Table 1, evidencing that Gd has been substituted for Zn to form substitutional compounds, for the ionic radius of Gd<sup>3+</sup> is 1.05 Å, which is larger than that (0.74 Å) of Zn<sup>2+</sup>, leading to expansion of the host lattice.

The electrical resistivity and thermopower *versus* temperature for  $\beta$ -(Zn<sub>1-x</sub>Gd<sub>x</sub>)<sub>4</sub>Sb<sub>3</sub> ( $x = 0, 0.001, 0.002$  and  $0.003$ ) samples in the temperature range of 300–655 K are depicted in Fig. 1. One can see from Fig. 1(a) that the temperature dependence of  $\rho$  for all samples is similar: it initially increases with increasing temperature and then decreases with further increase in temperature. This reduction of  $\rho$  at high temperatures can be ascribed to thermal excitation of minority carriers. Moreover,  $\rho$  decreases with increasing Gd content (except for the sample with  $x = 0.003$ ). Especially, the resistivity of (Zn<sub>0.998</sub>Gd<sub>0.002</sub>)<sub>4</sub>Sb<sub>3</sub> decreases to 18.8  $\mu\Omega\text{m}$  at 300 K, which is 45% lower than that (34.2  $\mu\Omega\text{m}$ ) of the un-doped sample. The highest  $\rho$  value for the un-doped sample appears at  $T_p \sim 525$  K, while this peak temperature  $T_p$  shifts to higher temperature for the doped samples, particularly,  $T_p$  appears at 575 K for  $x = 0.002$ , suggesting that Gd doping can inhibit the thermal excitation of minority carriers to some degree.

In contrast, thermopower  $S$  for compounds does not change obviously with the Gd content in the whole temperature range investigated (Fig. 1(b)). The positive values of  $S$  indicate that the major charge carriers are holes in all samples. The un-doped and doped samples have  $S \approx 120 \mu\text{V K}^{-1}$  at RT and almost increase linearly to 180  $\mu\text{V K}^{-1}$  with increasing temperature below  $\sim 575$  K, and then show weak temperature dependence with small reduction with further increasing temperature.

The positive Hall coefficient  $R_H$  for  $\beta$ -(Zn<sub>1-x</sub>Gd<sub>x</sub>)<sub>4</sub>Sb<sub>3</sub> ( $x = 0, 0.001, 0.002$  and  $0.003$ ) indicates hole conduction in this

**Table 1** List of room temperature lattice constants ( $a$  and  $c$ ), carrier concentration ( $p$ ), carrier mobility ( $\mu$ ), and the Lorenz number  $L$  of the bulk samples for  $\beta$ -(Zn<sub>1-x</sub>Gd<sub>x</sub>)<sub>4</sub>Sb<sub>3</sub> ( $x = 0, 0.001, 0.002$ , and  $0.003$ ) compounds

$x$	$a^a$ (Å)	$c^a$ (Å)	$P^b$ ( $10^{19} \text{ cm}^{-3}$ )	$\mu^c$ ( $\text{cm}^{-2} \text{ V}^{-1} \text{ s}^{-1}$ )	$L^d$ ( $\times 10^{-8} \text{ V}^2 \text{ K}^{-2}$ )
$x = 0$	12.218	12.411	8.0	22.9	1.48
$x = 0.001$	12.217	12.420	11.4	19.8	1.48
$x = 0.002$	12.223	12.421	17.6	18.9	1.49
$x = 0.003$	12.226	12.426	14.0	18.5	1.48

<sup>a</sup>  $a$  and  $c$  are the lattice parameters. <sup>b</sup>  $p$  is the carrier concentration. <sup>c</sup>  $\mu$  is the Hall mobility. <sup>d</sup>  $L$  is the Lorenz number.

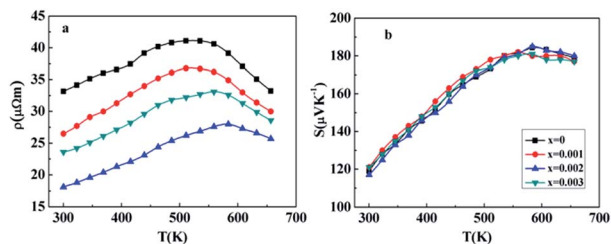


Fig. 1 Temperature dependence of (a) electrical resistivity and (b) thermopower for  $\beta\text{-(Zn}_{1-x}\text{Gd}_x)_4\text{Sb}_3$  ( $x = 0, 0.001, 0.002$  and  $0.003$ ) compounds.

system, being consistent with the results of  $S$  data. Assuming parabolic bands and a single band conduction process at 300 K, the obtained hole concentration  $p$  is  $8.0 \times 10^{19}$ ,  $11.4 \times 10^{19}$ ,  $17.6 \times 10^{19}$ , and  $14.0 \times 10^{19} \text{ cm}^{-3}$  as  $x$  increases from 0 to 0.001, 0.002, and then 0.003, respectively, as shown in Table 1. As we know,  $\beta\text{-Zn}_4\text{Sb}_3$  is a p-type semiconductor and has two kinds of Zn sites in its framework: lattice Zn sites (among which Zn occupancy is  $\sim 90\%$ ) and interstitial Zn sites.<sup>30</sup> In other words, its carriers (holes) come from vacancies in Zn lattice sites. Gd-doping will affect the vacancy content, which eventually leads to the changes of the carrier concentration: the more the vacancies, the more the holes. Hence, when the Gd content  $x = 0.001$  and  $0.002$ , the carrier concentration increases, indicating that the number of Zn vacancies increases, and this increased vacancies can scatter phonon, leading to decrease of the lattice thermal conductivity (see Fig. S2†). However, for the sample with  $x = 0.003$ , the hole concentration decreases, which indicates that the number of Zn vacancies decreases.

The measurements of the carrier concentration indicate that the decrease in resistivity  $\rho$  for the doped samples originates from the increase in the carrier concentration. An interesting phenomenon one notices here is that there is no reduction in  $S$  (seen in Fig. 1(b)) for doped samples though  $p$  increases, which seems to be in conflict with the Mott equation (eqn (1)), which implies other physics mechanism works. By using a single parabolic band model, the density state effective mass  $m_d^*$  and  $S$  can be expressed as:<sup>6</sup>

$$m_d^* = \frac{h^2}{2k_B T} \left( \frac{p}{4\pi F_{1/2}(\xi_F)} \right)^{2/3}, \quad (2)$$

$$S = \frac{k_B}{e} \left[ \frac{\left[ \left( r + \frac{5}{2} \right) F_{r+\frac{5}{2}}(\xi_F) \right]}{\left[ \left( r + \frac{3}{2} \right) F_{r+\frac{3}{2}}(\xi_F) \right]} - \xi_F \right] \quad (3)$$

$$= \frac{k_B}{e} \left[ \frac{2F_1(\xi_F)}{F_0(\xi_F)} - \xi_F \right],$$

with the Fermi integral of order  $i$ ,

$$F_i(\xi_F) = \int_0^\infty \frac{x^i}{1 + e^{(x-\xi_F)}} dx. \quad (4)$$

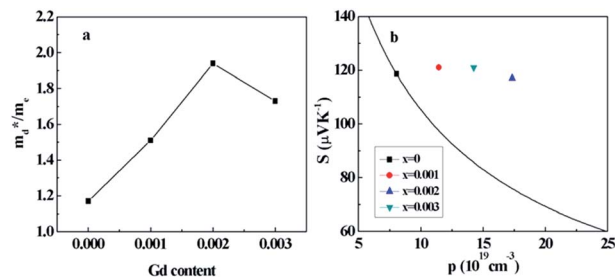


Fig. 2 (a) The ratio of density of states effective mass  $m_d^*$  to that of the free electron at 300 K, (b) variation of thermopower with the carrier concentration for  $\beta\text{-(Zn}_{1-x}\text{Gd}_x)_4\text{Sb}_3$  ( $x = 0, 0.001, 0.002$  and  $0.003$ ) compounds. The solid line is the carrier concentration dependence of thermopower for the un-doped  $\beta\text{-Zn}_4\text{Sb}_3$ , which is calculated by using formulae (2) and (3) for  $m_d^* = 1.17m_e$ .

Here  $k_B$  is the Boltzmann constant,  $h$  is the Plank constant, and  $\xi_F$  is the reduced Fermi level ( $E_f/k_B T$ ). In our calculations, we assume acoustic phonon scattering dominates (*i.e.*,  $r = -1/2$ ) in the compounds and uses the experimental data of the  $p$  and  $S$ . The density of states effective mass  $m_d^*$  for all the samples can be calculated and the obtained effective mass  $m_d^*/m_e$  (where  $m_e$  is the free electron mass) at 300 K is shown in Fig. 2(a).  $m_d^*$  of  $\beta\text{-(Zn}_{1-x}\text{Gd}_x)_4\text{Sb}_3$  reaches  $1.51m_e$ ,  $1.94m_e$ , and  $1.73m_e$  for  $x = 0.001, 0.002$  and  $0.003$ , which is 1.3, 1.7 and 1.5 times larger than that of the un-doped sample, respectively (it should be pointed out that the calculated effective mass  $m_d^*$  (at 300 K) for un-doped  $\beta\text{-Zn}_4\text{Sb}_3$  is  $1.17m_e$  that agrees well with the result ( $1.18m_e$ ) reported by Caillat *et al.*<sup>6</sup>). This increase in  $m_d^*$  actually signifies the increase in DOS. That is, the increase in  $m_d^*$  implies a non-parabolic perturbation in the electron dispersion relations, *i.e.* resonant distortion of DOS of  $\beta\text{-Zn}_4\text{Sb}_3$  near the Fermi level. In fact, Gd doping results in  $41 \mu\text{V K}^{-1}$  and  $36 \mu\text{V K}^{-1}$  increase in thermopower at 300 K for the doped compounds with  $x = 0.002$  and  $0.003$ , respectively, as shown in Fig. 2(b), where the solid line shows the dependence of  $S$  on the carrier concentration calculated using formulae (2) and (3) and  $m_d^* = 1.17m_e$  for  $\beta\text{-Zn}_4\text{Sb}_3$ , indicating that all the data of  $S$  with different Gd contents would fall on this line if there were no other mechanism (*i.e.* local resonant distortion of DOS) functions upon Gd doping.

To verify resonance distortion of DOS occurring in the doped compounds  $\beta\text{-(Zn}_{1-x}\text{Gd}_x)_4\text{Sb}_3$  ( $x \neq 0$ ), low temperature ( $< 4$  K) heat capacity  $C_p$  was measured for both un-doped  $\beta\text{-Zn}_4\text{Sb}_3$  and the doped compounds. As is well known, low temperature ( $< 4$  K) heat capacity  $C_p$  for a solid has a temperature dependence:  $C_p = \gamma T + bT^3$ , in which the term  $bT^3$  stands for the lattice contribution and  $\gamma T$  the contribution from charge carriers with  $\gamma$  being related to  $N(E_f)$  (here  $N(E_f)$  is electronic DOS at the Fermi level):

$$\gamma = \frac{\pi^2}{3} k_B^2 N(E_f). \quad (5)$$

However, in the Gd-doped samples magnetic heat capacity  $C_m$  from 4f electrons of Gd cannot be ignored. Then, the total

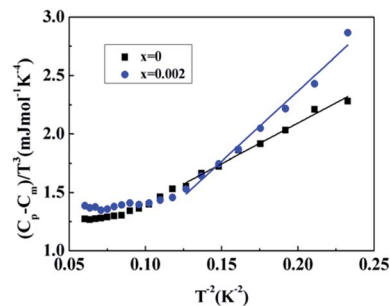


Fig. 3 The plots of  $(C_p - C_m)/T^3$  vs.  $T^{-2}$  for  $\beta$ -( $Zn_{1-x}Gd_x$ ) $_4Sb_3$  ( $x = 0$  and  $0.002$ ) compounds.

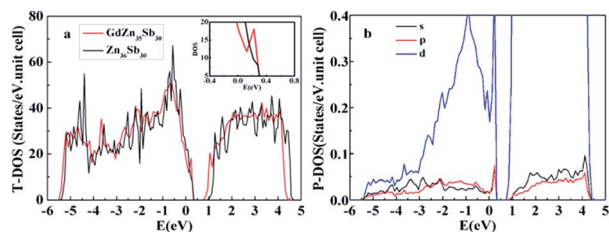


Fig. 4 (a) The total DOS (T-DOS) of  $Zn_{36}Sb_{30}$  and  $GdZn_{35}Sb_{30}$ , and (b) Gd partial DOS (P-DOS). The energy is in respect to the host valence band maximum.

heat capacity can be written as:  $C_p = \gamma T + bT^3 + C_m$ . According to the work of M. J. Parsons *et al.*,<sup>35,36</sup> magnetic heat capacity  $C_m$  for the Gd-doped sample with  $x = 0.002$ , at  $T < 9$  K, can be derived as  $C_m = 3.6T$  ( $mJ mol^{-1} K^{-1}$ ) (see ESI†). By subtracting  $C_m$  from  $C_p$  one has  $C_p - C_m = \gamma T + bT^3$  for the Gd-doped samples. Hence, the slope of a plot  $C_p/T^3$  (or  $(C_p - C_m)/T^3$ ) vs.  $1/T^2$  gives  $\gamma$  that reflects directly values of DOS at the Fermi level. Fig. 3 shows the plots of  $C_p/T^3$  vs.  $1/T^2$  for un-doped  $\beta$ - $Zn_4Sb_3$  and a typical doped compound  $\beta$ -( $Zn_{1-x}Gd_x$ ) $_4Sb_3$  ( $x = 0.002$ ) (where  $(C_p - C_m)/T^3$  vs.  $1/T^2$  is plotted). One can see that the slope ( $\gamma$ ) of the plot  $(C_p - C_m)/T^3$  vs.  $1/T^2$  for the doped compound  $\beta$ -( $Zn_{0.998}Gd_{0.002}$ ) $_4Sb_3$  is substantially larger than that for the un-doped one. By linear fitting of the curves in the low temperature regime, one obtains the ratio  $\gamma_{dop}/\gamma_{un-dop} = N(E_F)_{dop}/N(E_F)_{un-dop} = \sim 1.91$ , which is in good agreement with the ratio of density of states effective mass (see Fig. 2(a)), revealing that Gd doping indeed causes great increase in DOS near the Fermi level of the compound.

Our experimental observations are further confirmed by our theoretical calculations (Fig. 4). One can see from Fig. 4(a) that as compared with DOS of the pristine  $\beta$ - $Zn_4Sb_3$ , there is a sharp peak appearing in the DOS and near the edge of the valence-band for the Gd doping  $\beta$ - $Zn_4Sb_3$  (see the inset in Fig. 4(a)). Partial DOS analysis indicates that the resonant distortion of DOS of the Gd doped  $\beta$ - $Zn_4Sb_3$  comes mainly from contribution of the d-state of Gd (Fig. 4(b)), other than small contribution from p- and s-states. The present result indicates clearly that Gd doping causes resonant distortion of DOS of  $\beta$ - $Zn_4Sb_3$ , which will result in the enhancement of  $m^*$  and thermopower, which is in good agreement with our experimental result.

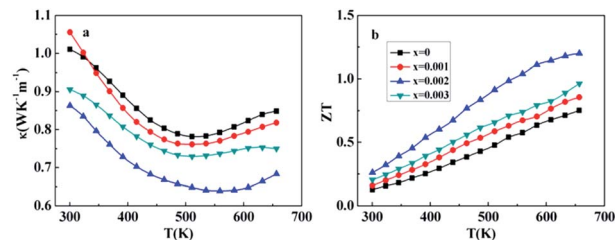


Fig. 5 Temperature dependence of (a) total thermal conductivity  $\kappa$  and (b)  $ZT$  for  $\beta$ -( $Zn_{1-x}Gd_x$ ) $_4Sb_3$  ( $x = 0, 0.001, 0.002$  and  $0.003$ ) compounds.

The temperature behavior of total thermal conductivity for ( $Zn_{1-x}Gd_x$ ) $_4Sb_3$  ( $x = 0, 0.001, 0.002$  and  $0.003$ ) is plotted in Fig. 5(a). In addition to the enhancement of thermopower through resonant distortion of DOS, Gd doping also causes a substantial decrease in  $\kappa$  in the measured temperature range. Thermal conductivity for all the samples initially decreases with increasing temperature and then increases with further increasing temperature. At 300 K, the sample with  $x = 0.002$  and  $0.003$  reduces to  $0.86 W m^{-1} K^{-1}$  and  $0.91 W m^{-1} K^{-1}$ , respectively, which is  $\sim 15\%$  and  $\sim 10\%$  smaller than that ( $1.01 W m^{-1} K^{-1}$ ) of the un-doped one. To determine the electronic component of the thermal conductivity, the Lorenz number  $L$  was estimated using formula (6) with the assumption of transport dominated by acoustic scattering and a single parabolic band.<sup>37</sup>

$$L = \left(\frac{k_B}{e}\right)^2 \frac{3F_0(\xi_F)F_2(\xi_F) - 4F_1(\xi_F)^2}{F_0(\xi_F)^2} \quad (6)$$

The obtained values of  $L$  are  $1.48$ – $1.49 \times 10^{-8} V^2 K^{-2}$ , as listed in Table 1. Therefore,  $\kappa_L$  can be obtained by subtracting  $\kappa_c$  ( $\kappa = \kappa_L + \kappa_c$ ). As shown in Fig. S2,†  $\kappa_L$  decreases with increasing doping content of Gd (except  $x = 0.003$ ) due to phonon scattering of impurity (Gd) and increased Zn vacancies. As mentioned above, the hole concentration decreases for the sample with  $x = 0.003$ , which suggests that the number of Zn vacancies decreases. Hence, the abnormal increase of  $\kappa$  for the sample with  $x = 0.003$  as compared to that for  $x = 0.002$  could be caused by the reduced Zn vacancies in lattice sites, which should cause an increase of the lattice conductivity due to weakened phonon scattering.

Fig. 5(b) gives the dimensionless figure of merit,  $ZT$ , for all samples as functions of temperature. The behavior of  $ZT$  for the four samples is similar: it increases with increasing temperature, and reaches the maximum value at 655 K. As a result of enhancement of thermopower and reduction in thermal conductivity, the largest  $ZT$  of 1.2 is achieved for  $\beta$ -( $Zn_{0.998}Gd_{0.002}$ ) $_4Sb_3$ , which is  $\sim 60\%$  larger than the un-doped sample ( $ZT = 0.75$ ).

## Conclusions

In summary, our experimental studies indicate that Gd-doping causes resonant distortion of DOS near the Fermi level of  $\beta$ -( $Zn_{1-x}Gd_x$ ) $_4Sb_3$ , which is manifested by a large increase in DOS effective mass and verified by low-temperature heat



capacity. First-principles calculations further reveal that a high sharp resonant peak in the DOS locating near the valence band maximum of  $\beta$ - $\text{Zn}_4\text{Sb}_3$  originates largely from contribution of the d-orbit of Gd. This resonant distortion of DOS results in an increase of thermopower by  $\sim 40 \mu\text{V K}^{-1}$  for  $\beta$ - $(\text{Zn}_{1-x}\text{Gd}_x)_4\text{Sb}_3$  ( $x = 0.002$  and  $0.003$ ); additionally, this Gd-doping gives rise to  $\sim 15\%$  reduction of thermal conductivity  $\kappa$  at  $x = 0.002$  content. As a result, a largest value of  $ZT = 1.2$  is achieved at 655 K for  $\beta$ - $(\text{Zn}_{1-x}\text{Gd}_x)_4\text{Sb}_3$  ( $x = 0.002$ ), which is  $\sim 60\%$  larger than that ( $ZT = 0.75$ ) of the un-doped one. The present result demonstrates that Gd doping is a promising way to elevate thermoelectric performance of  $\beta$ - $\text{Zn}_4\text{Sb}_3$  via bringing about resonant distortion of DOS.

## Author contribution

Baojin Ren synthesized the samples, measured the property and analysed data. Mian Liu contributed to the first-principles calculations. Xiaoguang Li was responsible for measuring the low-temperature heat capacity. Xiaoying Qin designed the experiments and analysed data. Di Li and Tianhua Zou helped in the measurements of thermoelectric properties. Guolong Sun and Jian Zhang contributed to microstructural characterization. Yuanyue Li and Hongxing Xin contributed to the synthesis of compounds.

## Acknowledgements

Financial support from the National Natural Science Foundation of China (nos 11374306, 11174292, 51101150, 50972146, and 10904144) and National Basic Research Program (973) of China (nos 2012CB922003 and 2015CB921201) are gratefully acknowledged.

## Notes and references

- 1 B. C. Sales, D. Mandrus and R. K. Williams, *Science*, 1996, **272**, 1325–1328.
- 2 G. Mahan, B. Sales and J. Sharp, *Phys. Today*, 1997, **50**, 42–47.
- 3 T. C. Harman, P. J. Taylor, M. P. Walsh and B. E. LaForge, *Science*, 2002, **297**, 2229–2232.
- 4 G. Chen, M. S. Dresselhaus, G. Dresselhaus, J. P. Fleurial and T. Caillat, *Int. Mater. Rev.*, 2003, **48**, 45–66.
- 5 G. J. Snyder and E. S. Toberer, *Nat. Mater.*, 2008, **7**, 105–114.
- 6 T. Caillat, J. P. Fleurial and A. Borshchevsky, *J. Phys. Chem. Solids*, 1997, **58**, 1119–1125.
- 7 T. Caillat and J. P. Fleurial, *Proc IECEC*, 1996, 905–909.
- 8 J. L. Cui, H. Fu, D. Y. Chen, L. D. Mao, X. L. Liu and W. Yang, *Mater. Charact.*, 2009, **60**, 824–828.
- 9 J. L. Cui, L. D. Mao, D. Y. Chen, X. Qian, X. L. Liu and W. Yang, *Curr. Appl. Phys.*, 2009, **9**, 713–716.
- 10 T. Koyanagi, K. Hino, Y. Nagamoto, H. Yoshitake and K. Kishimoto, *Proceedings Ict'97-Xvi International Conference on Thermoelectrics*, 1997, pp. 463–466.
- 11 D. Li, H. H. Hng, J. Ma and X. Y. Qin, *J. Mater. Res.*, 2009, **24**, 430–435.
- 12 A. P. Litvinchuk, J. Nylén, B. Lorenz, A. M. Guloy and U. Häussermann, *J. Appl. Phys.*, 2008, **103**, 123524.
- 13 F. Liu, X. Y. Qin and H. X. Xin, *J. Phys. D: Appl. Phys.*, 2007, **40**, 7811–7816.
- 14 G. Nakamoto, T. Souma, M. Yamaba and M. Kurisu, *J. Alloys Compd.*, 2004, **377**, 59–65.
- 15 B. L. Pedersen, H. Birkedal, E. Nishibori, A. Bentien, M. Sakata, M. Nygren, P. T. Frederiksen and B. B. Iversen, *Chem. Mater.*, 2007, **19**, 6304–6311.
- 16 S.-C. Ur, I.-H. Kim and P. Nash, *Mater. Lett.*, 2004, **58**, 2132–2136.
- 17 B. L. Pedersen, H. Birkedal, M. Nygren, P. T. Frederiksen and B. B. Iversen, *J. Appl. Phys.*, 2009, 105.
- 18 B. L. Pedersen, H. Yin, H. Birkedal, M. Nygren and B. B. Iversen, *Chem. Mater.*, 2010, **22**, 2375–2383.
- 19 S. Y. Wang, X. J. Tan, G. J. Tan, X. Y. She, W. Liu, H. Li, H. J. Liu and X. F. Tang, *J. Mater. Chem.*, 2012, **22**, 13977–13985.
- 20 J. Nylen, S. Lidin, M. Andersson, B. B. Iversen, H. X. Liu, N. Newman and U. Haussermann, *Chem. Mater.*, 2007, **19**, 834–838.
- 21 J. P. Heremans, V. Jovovic, E. S. Toberer, A. Saramat, K. Kurosaki, A. Charoenphakdee, S. Yamanaka and G. J. Snyder, *Science*, 2008, **321**, 554–557.
- 22 G. D. Mahan and J. O. Sofo, *Proc. Natl. Acad. Sci. U. S. A.*, 1996, **93**, 7436–7439.
- 23 M. Cutler and N. F. Mott, *Phys. Rev.*, 1969, **181**, 1336–1340.
- 24 Q. Q. Wang, X. Y. Qin, D. Li, R. R. Sun, T. H. Zou and N. N. Wang, *J. Appl. Phys.*, 2013, **113**, 124901.
- 25 Q. Q. Wang, X. Y. Qin, D. Li and T. H. Zou, *Appl. Phys. Lett.*, 2013, **102**, 154101.
- 26 M. Liu, X. Qin, C. Liu, X. Li and X. Yang, *J. Alloys Compd.*, 2014, **584**, 244–248.
- 27 A. S. Mikhaylushkin, J. Nylen and U. Haussermann, *Chemistry*, 2005, **11**, 4912–4920.
- 28 J. Nylen, M. Andersson, S. Lidin and U. Haussermann, *J. Am. Chem. Soc.*, 2004, **126**, 16306–16307.
- 29 Y. A. Ugai, E. M. Averbakh and V. V. Lavrov, *Phys. Solid State*, 1963, **4**, 2393–2395.
- 30 F. Cargnoni, E. Nishibori, P. Rabiller, L. Bertini, G. J. Snyder, M. Christensen, C. Gatti and B. B. Iversen, *Chemistry*, 2004, **10**, 3861–3870.
- 31 G. J. Snyder, M. Christensen, E. Nishibori, T. Caillat and B. B. Iversen, *Nat. Mater.*, 2004, **3**, 458–463.
- 32 P. E. Blochl, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 1994, **50**, 17953–17979.
- 33 G. Kresse and D. Joubert, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 1999, **59**, 1758–1775.
- 34 J. P. Perdew, K. Burke and M. Ernzerhof, *Phys. Rev. Lett.*, 1996, **77**, 3865–3868.
- 35 M. J. Parsons, J. Crangle, B. Dennis, K. U. Neumann and K. R. A. Ziebeck, *Czech J. Phys.*, 1996, **46**, 2057–2058.
- 36 M. J. Parsons, J. Crangle, K. U. Neumann and K. R. A. Ziebeck, *J. Magn. Magn. Mater.*, 1998, **184**, 184–192.
- 37 E. S. Toberer, P. Rauwel, S. Gariel, J. Tafto and G. J. Snyder, *J. Mater. Chem.*, 2010, **20**, 9877–9885.